

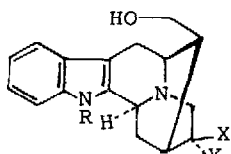
TOTAL SYNTHESIS OF N<sub>a</sub>-METHYL- $\Delta^1\delta$ -ISOKOUMIDINE, A POSSIBLE  
PRECURSOR OF THE KOUMINE TYPE INDOLE ALKALOIDS [1]

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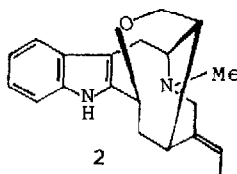
**Abstract** A novel total synthesis of title compound 3 is described.

The recent appearance of a communication from Magnus et al. on the total synthesis of (+)koumidine 1 and (+)taberpsychine 2 [2] prompts us to report our recent result on the total synthesis of the title compound 3. Based on the transformation of (-)taberpsychine to natural (-)koumine 4 [3] and ready access of the former from koumidine [2,4], 3 might be considered as a key intermediate for the total synthesis of N<sub>a</sub>-methyl koumine or, after demethylation, of koumine itself, inspite of the presence of a vinyl side chain in 3 instead of an ethylidene side chain in 1.

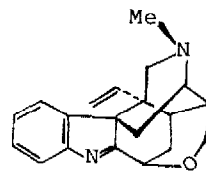


1 X, Y: =CHCH<sub>3</sub> (Z), R: H

3 X: H, Y: CH=CH<sub>2</sub>, R: Me



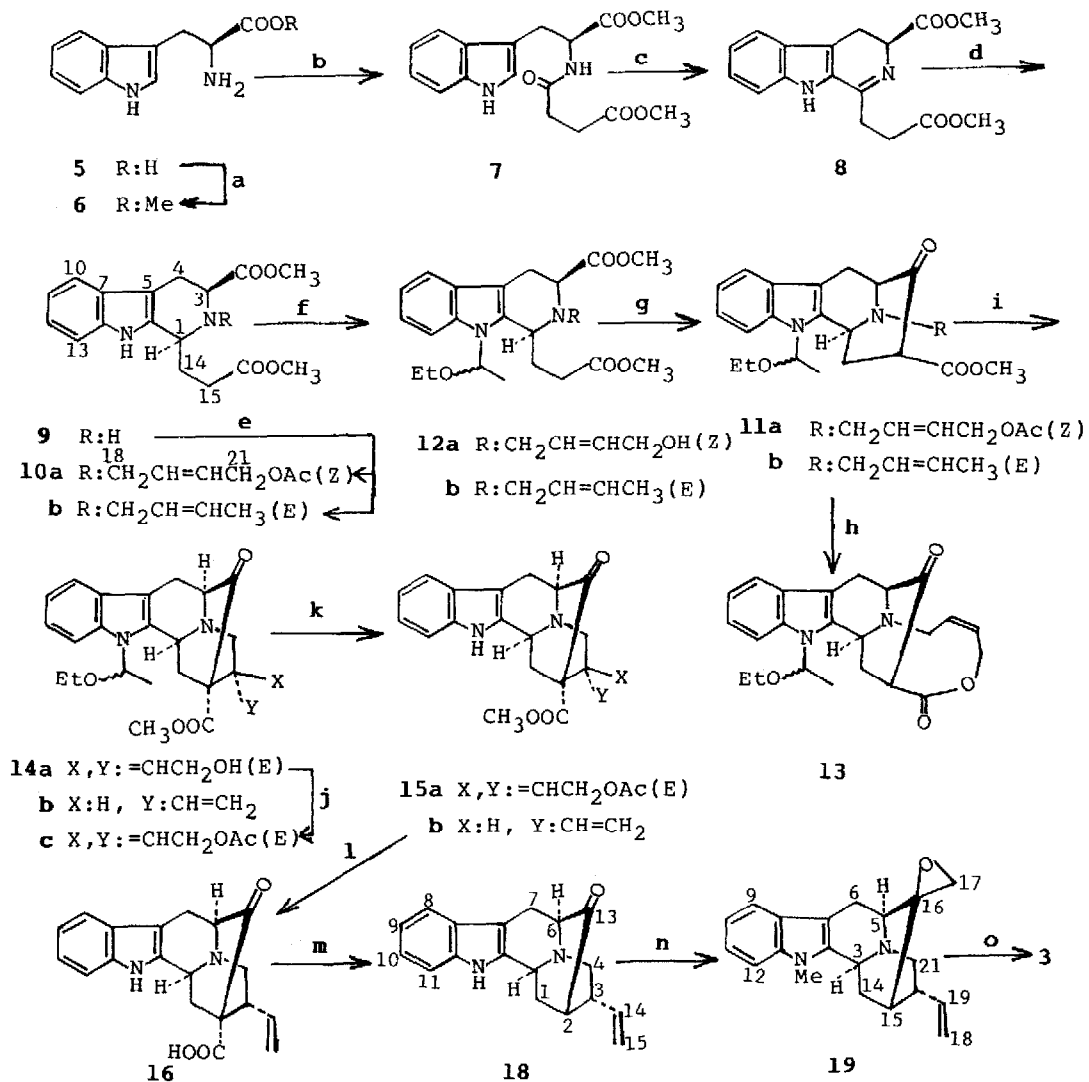
2



4

With our recently developed methodology of the asymmetric synthesis of 1,2,3,4-tetrahydro- $\beta$ -carboline in hand [5], L-tryptophan 5 was used as the starting material for our approach. Treatment of the methyl ester of L-tryptophan 6, which was obtained by exposure of L-tryptophan 5 to SOCl<sub>2</sub>/MeOH/r.t./2d [6], with succinic half methyl ester acid chloride (Py/r.t./2d) afforded amide 7 in 81% yield, [ $\alpha$ ]<sub>D</sub>+80° (c1.1 in CHCl<sub>3</sub>). The labile  $\beta$ -carboline 9 was obtained through a Bischler-Napieralski reaction (POCl<sub>3</sub>/60°C/10h), followed by hydrogenolysis of the resulting 8 (PtO<sub>2</sub>/EtOH) in 99% yield [7]. N<sub>b</sub>-alkylation of 9 to give 10 was achieved by treatment with 1-bromo-4-acetoxy-2-butene-2 or 1-bromo-E-butene-2/NaHCO<sub>3</sub>/CH<sub>3</sub>CN. Exposure of 10 to CH<sub>2</sub>=CHOEt/PTS (cat.)/30°C afforded the N<sub>a</sub>-protected tetrahydro- $\beta$ -carboline 11. Then the Dieckmann cyclizations gave the expected  $\beta$ -keto esters 12.

After the failure to complete the desired sarpagine ring system via a



**Reagents and Conditions:** a: SOCl<sub>2</sub>, MeOH, rt, 2d. b: MeOOCCH<sub>2</sub>CH<sub>2</sub>COCl, Py, rt, 2d. c: POCl<sub>3</sub>, 60°C, 10h. d: H<sub>2</sub>, PtO<sub>2</sub>; EtOH. e: BrCH<sub>2</sub>CH=CHCH<sub>2</sub>OAc(Z) or BrCH<sub>2</sub>CH=CHCH<sub>3</sub>(E), NaHCO<sub>3</sub>, CH<sub>3</sub>CN, reflux, 10h. f: CH<sub>2</sub>=CHOEt, PTS(cat.), CH<sub>2</sub>Cl<sub>2</sub>, 30°C, 10h. g: 11a→12a: NaH, MeOH(cat.), C<sub>6</sub>H<sub>6</sub>, reflux, 6h. 11b→12b: (Me<sub>3</sub>Si)<sub>2</sub>NNa, DME, 100°C, 5.5h. h: DCC, CF<sub>3</sub>COOH, DMSO, C<sub>6</sub>H<sub>6</sub>, Py, rt, 8h. i: Mn(OAc)<sub>3</sub>·H<sub>2</sub>O, Cu(OAc)<sub>2</sub>·H<sub>2</sub>O, HOAc, rt, 8h. j: Ac<sub>2</sub>O, Py, rt, 24h. k: HOAc-MeOH-H<sub>2</sub>O(2:1:1), 90°C, 2h. l: 2N KOH, MeOH, 60°C, 2h. m: N-hydroxy-2-pyridinethione 17, DCC, DMAP, THF or CH<sub>2</sub>Cl<sub>2</sub>, rt, 2d; then t-BuSH, hv, 2-3h. n: NaH, DMSO; then (CH<sub>3</sub>)<sub>3</sub>Si, THF, rt, 20h. o: AlClH<sub>2</sub>, THF, reflux, 5h.

$\text{Pd}^0$  mediated alkylation of **12a** ( $\text{Pd}(\text{PPh}_3)_4$ ,  $\text{As}_2\text{O}_3$ ), we attempted to oxidize **12a** with DCC/DMSO/ $\text{CF}_3\text{COOH}$  into the corresponding  $\alpha,\beta$ -unsaturated aldehyde in order to conduct a subsequent intramolecular Michael addition. Quite unexpectedly, **12a** preferred an intramolecular ester exchange to form the 10-membered macrolacton **13** instead of undergoing the desired oxidation into the corresponding aldehyde.

We then turned to apply the recently developed cyclization process of alkenes<sup>[9]</sup> which might be more efficient in strained molecules. It involved a manganese(III)-based oxidative free radical cyclization, which was initiated by oxidation of  $\beta$ -carbonyl compound with  $\text{Mn}(\text{OAc})_3 \cdot \text{H}_2\text{O}/\text{Cu}(\text{OAc})_2 \cdot \text{H}_2\text{O}$  in situ. Fortunately, in our case, treatment of **12** with  $\text{Mn}(\text{OAc})_3 \cdot \text{H}_2\text{O}/\text{Cu}(\text{OAc})_2 \cdot \text{H}_2\text{O}$  in HOAc readily effected the expected cyclization in almost quantitative yield.

After removal of the  $\text{N}_a$ -protecting group in **14b**, **14c** by heating in HOAc: MeOH: $\text{H}_2\text{O}$ =2:1:1, **15a** and **15b** were, respectively, obtained in good yield.

Many attempts to remove the bridge-head carboxylic group by conventional methods<sup>[12]</sup>, have all failed. However, we finally found that Barton's decarboxylation method<sup>[11]</sup> proved effective in our case. Treatment of **16** with DCC/DMAP/**17** in excess dry THF or  $\text{CH}_2\text{Cl}_2$ , followed by the addition of *t*-BuSH and photolysis with a 300W tungsten slide lamp for several hours afforded **18** in 55% yield,  $[\alpha]_D^{+11.7}$  (c0.6 in  $\text{CHCl}_3$ )<sup>[10]</sup>.

At this stage all what we needed to do was to homologate the carbonyl group of **18** into a hydroxymethyl function in a stereospecific manner. While this carbonyl appeared to resist conventional Wittig reagents, treatment of **18** with  $(\text{CH}_3)_2\text{S}=\text{CH}_2/\text{DMSO}/\text{THF}/\text{r.t.}$ , followed by reduction of the resulting epoxide **19** with  $\text{AlH}_2\text{Cl}/\text{THF}$  regio- and stereoselectively gave rise to  $\text{N}_a$ -methyl- $\Delta^{18}$ -iso-koumidine **3**<sup>[13]</sup>.

Since  $\text{N}_a$ -methyl ajmaline has been recently converted into koumidine by Sakai<sup>[4]</sup>, we believe that further elaboration of **3** would lead to  $\Delta^{18}$ -iso-koumidine and eventually to the total synthesis of koumine.

The key intermediate **18** might also be an important precursor for synthesis of some other interesting indole alkaloids, such as the gelesmine type, the vobasine type, and the humantenine type. Those synthetic endeavors are now in progress in our laboratory. The crucial intramolecular radical cyclization (**12**→**14**) would offer a powerful way for constructing quinuclidine skeleton which might be valuable in other areas of alkaloid synthesis<sup>[14]</sup>.

## References and Notes

1. A part of our results was reported on the "2nd Sino-French Symposium on the Chemistry of Natural Products", Sept. 20-22, 1988, La Londe les Maures, France.

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7. The reduction **8**→**9** yielded within the limits of HPTLC and n.m.r. detection a single epimer, of which the C-3 configuration was determined as S (C-3H cis to C-5H) by applying the known  $^{13}\text{C}$  NMR method<sup>[8]</sup> on its stable derivative **11a**.  $^{13}\text{C}$  NMR spectrum of **11a**:  $\delta(\text{CDCl}_3)$ : 20.94(CH<sub>3</sub>), 21.23(CH<sub>2</sub>, 4-C), 28.94(CH<sub>2</sub>, 15-C), 50.95(CH, 1-C), 51.64(OCH<sub>3</sub>), 51.93(OCH<sub>3</sub>), 55.64(CH, 3-C), 59.40(CH<sub>2</sub>, 18-C), 60.13(CH<sub>2</sub>, 21-C), 106.70(C, 5-C), 110.89(CH, 13-C), 118.11(CH, 10-C), 119.33(CH, 11-C), 121.67(CH, 12-C), 126.85(C, 7-C), 131.39(CH, 19-C), 133.14(CH, 20-C), 136.17(2C, 6,8-C), 170.82(C=O), 173.65(C=O), 174.63(C=O)ppm.
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10. **18**: IR(KCl): 1725cm<sup>-1</sup>.  $^1\text{H}$  NMR(CDCl<sub>3</sub>): 1.79(1H, m, 1-Hax), 2.32(2H, m, 1-Heq), 2.46(1H, dXd, J<sub>1</sub>=6Hz, J<sub>2</sub>=13Hz, 4-Hax), 2.68(1H, dXd, J<sub>1</sub>=8Hz, J<sub>2</sub>=13Hz, 4-Heq), 2.78(2H, m, 2,3-H), 2.94(1H, dXd, J<sub>1</sub>=6.2Hz, J<sub>2</sub>=15.5Hz, 7-Hax), 3.31(1H, d(d), J<sub>1</sub>=1Hz, J<sub>2</sub>=15.5Hz, 7-Heq), 3.46(1H, m, 12b-H), 3.70(1H, d(d), J<sub>1</sub>=1Hz, J<sub>2</sub>=6.2Hz, 6-H), 5.14(2H, m, 15-H), 5.62(1H, m, 14-H), 7.15(3H, m, 9,10,11-H), 7.49(1H, m, 8-H), 8.34(1H, s, 12-H)ppm.
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13. **3**: IR(CHCl<sub>3</sub>): 3400cm<sup>-1</sup>. MS: 308(M<sup>+</sup>), 290, 183(base peak).
14. All new compounds were characterized by  $^1\text{H}$  NMR,  $^{13}\text{C}$  NMR, IR, HRMS and/or microanalysis.  
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